Spirobacillenes A and B, Unusual Spiro-cyclopentenones from *Lysinibacillus fusiformis* KMC003

2012 Vol. 14, No. 19 5002–5005

ORGANIC LETTERS

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Received July 31, 2012





Two previously unreported spiro-cyclopentenones, spirobacillenes A (1) and B (2), were isolated from the 24 h broth culture of *Lysinibacillus fusiformis* KMC003 derived from acidic coal-mine drainage. The structures of 1 and 2 were elucidated by analyses of the NMR, HRFABMS, single-crystal X-ray diffraction crystallography, and circular dichroism (CD) spectral data. Compound 1 possessed moderate inhibitory activity against the production of nitric oxide (NO) and reactive oxygen species (ROS).

Microbial metabolites have provided a variety of bioactive chemical structures in drug discovery programs, and the chemical diversity from microbial resources is continuously increasing.^{1,2} Multidirectional approaches, such as exploitation of new species, have been developed for discovering novel chemical scaffolds from microorganisms. One effective approach for new secondary metabolites is to study microorganisms derived from unexplored extreme environments (e.g., acidic mine drainage), where microorganisms can create a unique offensive and defensive biochemical metabolism under ecological pressure.³ Recently, diverse microbes derived from acidic mine environments, which can grow and produce new secondary

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(2) Pettit, R. K. Appl. Microbiol. Biotechnol. 2009, 83, 19–25.

metabolites under laboratory conditions, have been described in the literature.⁴ The most prominent example is a *Penicillium* fungal strain isolated from the Berkeley Pit, which is an acid mine waste lake that is extremely contaminated by heavy metals and sulfuric acid. Structurally remarkable bioactive compounds were isolated from the *Penicillium* sp.⁵

Recently, we isolated a bacterial strain, *Lysinibacillus fusiformis* KMC003,⁶ from acidic coal mine drainage that

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⁽¹⁾ Corley, D. G.; Durley, R. C. J. Nat. Prod. **1994**, 57, 1484–1490.

^{(3) (}a) Rothschild, L. J.; Mancinelli, R. L. *Nature* **2001**, 409, 1092–1101. (b) Johnson, D. B. *FEMS Microbiol. Ecol.* **1998**, 27, 307–317. (c) Pettit, R. K. *Mar. Biotechnol.* **2011**, *13*, 1–11.

^{(4) (}a) Stierle, A. A.; Stierle, D. B.; Goldstein, E.; Parker, K.; Bugni, T.; Baarson, C.; Gress, J.; Blake, D. J. Nat. Prod. 2003, 66, 1097–1100.
(b) Stierle, D. B.; Stierle, A. A.; Hobbs, J. D.; Stokken, J.; Clardy, J. Org. Lett. 2004, 6, 1049–1052. (c) Stierle, A. A.; Stierle, D. B.; Kemp, K. J. Nat. Prod. 2004, 67, 1392–1395.

^{(5) (}a) Stierle, A. A.; Stierle, D. B.; Kelly, K. J. Org. Chem. 2006, 71, 5357–5360. (b) Stierle, D. B.; Stierle, A. A.; Patacini, B. J. Nat. Prod. 2007, 70, 1820–1823. (c) Stierle, A. A.; Stierle, D. B.; Patacini, B. J. Nat. Prod. 2008, 71, 856–860. (d) Stierle, D. B.; Stierle, A. A.; Patacini, B.; McIntyre, K.; Girtsman, T.; Bolstad, E. J. Nat. Prod. 2011, 74, 2273–2277.

⁽⁶⁾ The KMC-003 strain was classified as a *Lysinibacillus* sp. Grampositive bacterium based on a 16S rRNA sequence analysis. The strain shares 99.9% sequence identity with *Lysinibacillus fusiformis*.

was highly contaminated by iron-rich heavy-metal ions and sulfuric acid (pH 3.0). Lysinibacillus fusiformis, a Gram-positive bacterium, was reclassified via reinspection of rRNA group 2 of the Bacillus fusiformis and was found to have a distinctive cell-wall peptidoglycan composition.⁷ They were isolated from heavy-metal-contaminated wastewater and from the livers of the puffer fish.^{8,9} Previous studies of L. fusiformis reported their ability to produce tetrodotoxin and convert oleic acid to 10-hydroxystearic acid by oleate hydratase as well as their resistance to multiple metals.⁸⁻¹⁰ However, no bioactive secondary metabolites have been reported from the genus Lysinibacillus. Therefore, we investigated the production of secondary metabolites from the bacterial strain L. fusiformis KMC003 because the Bacillus species are known for their ability to produce structurally diverse bioactive molecules, such as polyene, macrolide, and especially peptide antibiotics.¹¹



A chemical analysis of the secondary metabolites produced by *L. fusiformis* KMC003 revealed the presence of six major metabolites in the 24 h broth culture. Interestingly, these metabolites disappeared after 48 h of culturing. To identify their chemical structures, the 24 h liquid culture of *L. fusiformis* KMC003 was extracted with ethyl acetate. The ethyl acetate extracts were subsequently subjected to reversed-phase HPLC separation to afford two novel compounds, spirobacillenes A (1) and B (2) along with two new natural compounds [(*Z*)-3-hydroxy-4-(3-indolyl)-1-hydroxyphenyl-2-butenone (3) and (*Z*)-3-hydroxy-4-(3-indolyl)-1-phenyl-2-butenone (4)] and two known indole derivatives [soraphinol A (5)¹² and kurasoin B (6)¹³].

(9) Wang, J.; Fan, Y.; Yao, Z. Toxicon 2010, 56, 640-643.

- (10) Kim, B.-N.; Joo, Y.-C.; Kim, Y.-S.; Kim, K.-R.; Oh, D.-K. Appl. Microbiol. Biotechnol. 2012, 95, 929–937.
- (11) Hamdache, A.; Lamarti, A.; Aleu, J.; Collado, I. G. J. Nat. Prod. **2011**, *74*, 893–899.

(12) (a) Li, X. M.; Zee, O. P.; Shin, H. J.; Seo, Y. W.; Ahn, J.-W. Bull. Korean Chem. Soc. **2007**, 28, 835–836. (b) Ahn, J.-W.; Li, X. M.; Zee, O.-P. Bull. Korean Chem. Soc. **2007**, 28, 1215–1216. Spirobacillenes A (1) and B (2) featured a unique indole and indolenine moiety that contained spiro-cyclopentenones, respectively. Herein, we report the isolation, structure elucidation, and possible biosynthetic pathway of 1 and 2.

Spirobacillene A $(1)^{14}$ was isolated as a yellow chunk crystal. The molecular formula was determined to be $C_{18}H_{13}NO_3$ ([M+H]⁺ at m/z 292.0977) based on a HRFABMS measurement, which indicated that 1 contained 13 degrees of unsaturation. The IR spectrum exhibited absorption bands that corresponded to hydroxyl (3228 cm^{-1}) and carbonyl (1693 cm^{-1}) functional groups. The ¹H NMR spectrum of **1** recorded in acetonitrile- d_3 displayed characteristics of a NH signal [$\delta_{\rm H}$ 9.63 (1H, br s)], three olefinic methine proton signals [$\delta_{\rm H}$ 7.46 (1H, s, overlap), 7.06 (2H, d, J = 10.0 Hz), 6.33 (2H, d, J = 10.0Hz)], four aromatic proton signals [$\delta_{\rm H}$ 7.98 (1H, d, J = 8.0Hz), 7.46 (1H, m, overlap), 7.23 (1H, ddd, J = 8.0, 7.0, 1.2Hz), 7.16 (1H, ddd, J = 8.0, 7.0, 1.2 Hz)], and a methylene group $[\delta_{\rm H} 2.76 (2H, s)]$. The ¹³C NMR spectrum indicated resonances for 16 carbons attributable to two α , β -unsaturated ketone carbons ($\delta_{\rm C}$ 197.6, 185.0), a methylene carbon $(\delta_{\rm C} 42.1)$, three olefinic methine carbons $(\delta_{\rm C} 154.2, 129.1,$ 125.3), four aromatic methines ($\delta_{\rm C}$ 122.7, 122.5, 120.1, and 111.7), five aromatic (or sp²) quaternary carbons ($\delta_{\rm C}$ 148.2, 137.0, 136.3, 126.0, and 109.2), and a quaternary carbon ($\delta_{\rm C}$ 47.0). The HSQC spectrum of 1 enabled the assignment of all of the protons to the directly bonded carbons. All of the above data suggested the presence of both an indole moiety and a cross-conjugated dienone group. The substructures were assigned by analyses of the $^{1}H-^{1}H$ COSY and $^{1}H-^{13}C$ HMBC spectral data. The ¹H⁻¹H COSY spectrum indicated connectivity between NH ($\delta_{\rm H}$ 9.63) and H-2 ($\delta_{\rm H}$ 7.46). The sequential COSY correlations from H-4 ($\delta_{\rm H}$ 7.98) to H-7 ($\delta_{\rm H}$ 7.46) and the HMBC correlations from H-5 ($\delta_{\rm H}$ 7.16) to C-9 ($\delta_{\rm C}$ 126.0), H-6 ($\delta_{\rm H}$ 7.23) to C-8 ($\delta_{\rm C}$ 136.3), and H-2 ($\delta_{\rm H}$ 7.46) to C-3 $(\delta_{\rm C} 109.2)$ and C-9 $(\delta_{\rm C} 126.0)$ established the presence of an indole structure. In addition, the COSY correlations of H-15/H-19 ($\delta_{\rm H}$ 7.06) and H-16/H-18 ($\delta_{\rm H}$ 6.33) and the observed HMBC correlations from both H-15/H-19 and H-16/H-18 to C-17 ($\delta_{\rm C}$ 185.0) and C-14 ($\delta_{\rm C}$ 47.0) established the presence of a 2,5-cyclohexadienone. This sixmembered ring contains a C-14 spiro-carbon center connected to a cyclopentenone through C-14, which was assigned based on HMBC correlations from H-15/H-19 $(\delta_{\rm H} 7.06)$ to C-13 $(\delta_{\rm C} 42.1)$ and C-10 $(\delta_{\rm C} 137.0)$. The additional HMBC correlations from H-13 ($\delta_{\rm H}$ 2.76) to C-10 (δ_C 137.0), C-11 (δ_C 148.2), C-12 (δ_C 197.6), C-14 $(\delta_{\rm C} 47.0)$, and C-15/C19 $(\delta_{\rm C} 154.2)$ led to the construction of a spiro[4.5]decane skeleton of 1.^{15,18} 2D NOESY experiments also supported the assignment of this spiro-[4.5]decane structure based on an NOE correlation of H-13

⁽⁷⁾ Ahmed, L.; Yokota, A.; Yamazoe, A.; Fujiwara, T. Int. J. Syst. Evol. Microbiol. 2007, 57, 1117–1125.

⁽⁸⁾ He, M.; Li, X.; Liu, H.; Miller, S. J.; Wang, G.; Rensing, C. J. Hazard. Mater. 2011, 185, 682–688.

^{(13) (}a) Uchida, R.; Shiomi, K.; Sunazuka, T.; Inokoshi, J.; Nishizawa, A.; Hirose, T.; Tanaka, H.; Iwai, Y.; Omura, S. J. Antibiot. **1996**, 49, 886–889. (b) Uchida, R.; Shiomi, K.; Inokoshi, J.; Masuma, R.; Kawakubo, T.; Tanaka, H.; Iwai, Y.; Omura, S. J. Antibiot. **1996**, 49, 932–934.

⁽¹⁴⁾ Spirobacillene A (1): yellow chunk crystal; mp 217–220 °C; $[\alpha]_D^{25}$ –4.4 (*c* 0.10, CH₃CN); UV (CH₃CN) λ_{max} (log ε) 342 (3.73), 224 (4.07), 193 (4.06) nm; IR (film) ν_{max} 3228, 1693, 1657, 1619, 1394, 1221, 858, 661 cm⁻¹; ¹H and ¹³C NMR spectra, see Table S2 in the SI; HRFABMS[M+H]⁺ *m/z* 292.0977 (calcd for C₁₈H₁₄NO₃, 292.0974).

⁽¹⁵⁾ Textor, A.; Papastavrou, I.; Siewert, J.; Magull, J.; Kulik, A.; Fiedler, H.-P.; von Zezschwitz, P.; Grond, S. *Chem.—Eur. J.* **2007**, *13*, 7416–7423.

and H-15/H-19. The structure of **1** has been unambiguously confirmed by single-crystal X-ray diffraction analysis as 3-hydroxy-4-(1*H*-indolyl)-spiro[4,5]deca-3,6,9-trienone (Figure 1).



Figure 1. X-ray ORTEP drawing of 1.

Spirobacillene B $(2)^{16}$ was obtained as an optically active pale-yellow oil ($[\alpha]_D^{25}$ -6.5) that possessed the same molecular formula, C₁₈H₁₃NO₃, as 1 based on the positive-ion HRFABMS (obsd $[M+H]^+$ at m/z 292.0978, calcd, 292.0974). The strong UV absorptions with λ_{max} values of 219 and 322 nm suggested the presence of an indole or phenyl conjugated α,β -unsaturated ketone chromophore in the structure of 2^{17} . The critical differences between the ¹H NMR signals of **1** and **2** were the absence of a NH ($\delta_{\rm H}$ 9.63) signal and the coupling constants (9.0 Hz) for H-15/ H-19 ($\delta_{\rm H}$ 7.10) and H-16/H-18 ($\delta_{\rm H}$ 6.65) in the ¹H NMR spectrum of 2, which suggested that 2 contained a hydroxyphenyl group instead of the cross-conjugated dienone group in 1.¹⁸ A hydroxylated aromatic carbon signal ($\delta_{\rm C}$ 158.0) and the chemical resonances of C-15/C-19 (δ_C 129.4) and C-16/C-18 ($\delta_{\rm C}$ 115.4) in the ¹³C NMR spectrum supported this result. The ${}^{13}C$ NMR spectral data of 2 also exhibited a nitrogen-bearing sp² carbon (or ester/acid carbonyl carbon) ($\delta_{\rm C}$ 177.2),¹⁹suggesting the presence of an indolenine core. The gross structure of 2 was constructed by analysis of the detailed ¹H-¹H COSY and HMBC spectral data (Figure 2). The COSY correlations from H-15/H-19 ($\delta_{\rm H}$ 7.10) and H-16/H-18 ($\delta_{\rm H}$ 6.65) and

the HMBC correlations from H-16/H-18 ($\delta_{\rm H}$ 6.65) to C-14 $(\delta_{\rm C} 124.6)$ and H-15/H-19 $(\delta_{\rm H} 7.10)$ to C-17 $(\delta_{\rm C} 158.0)$ led to the identification of a 1,4-disubstituted phenyl group. The linkage of the indolenine core and 1,4-disubstituted phenyl group was achieved based on the HMBC correlations from the methylene proton H-10 ($\delta_{\rm H}$ 2.92, 2.55) to C-2 ($\delta_{\rm C}$ 177.2), C-3 ($\delta_{\rm C}$ 61.4), C-9 ($\delta_{\rm C}$ 142.4), C-11 ($\delta_{\rm C}$ 199.0), and C-13 ($\delta_{\rm C}$ 135.8) and on the correlations from H-15/H-19 ($\delta_{\rm H}$ 7.10) to C-13 ($\delta_{\rm C}$ 135.8). Interestingly, these HMBC correlations indicated the presence of an α -hydroxy β -phenyl substituted cyclopentenone that was connected to the indolenine structure via spiro-carbon C-3 $(\delta_{\rm C} 61.4)$.^{19,20} The 2D NOESY experiments supported the existence of this spiro structure by NOE correlations between H-4 and H-10 β (or H-10 α), H-2 and H-10 α (or H-10 β), and H-2 and H-15/H-19. Therefore, the planar structure of 2 was assigned as 3-hydroxy-2-(4hydroxyphenyl)-spiro(cyclopentene-1,3'-indole)-4-one.



Figure 2. Key COSY and HMBC correlations of 1 and 2.

The absolute configuration at spiro carbon C-3 of 2 was determined by application of the exciton-coupled circular dichroism method (Figure 3). The CD spectrum of 2, which was acquired in acetonitrile, showed distinguishable absorption bands at 220 and 320 nm, similar to the UV spectrum. The Cotton effect was observed as a negative band at 223 nm and a positive band at 234 nm and is associated with the exciton coupling of the phenyl and indole chromophores.^{21,22} In addition, the CD spectrum exhibited a negative Cotton effect at 313 nm and a positive band at 324 nm, which could have originated from the exciton coupling between the phenyl conjugated cyclopentanone and indole chromophores. The completely split circular dichroism curve of 2 suggested that the helicity of the phenyl and indole chromophores was positive, which indicated that the absolute configuration of spirocarbon C-3 was $S^{22,23}$ Therefore, the structure of **2** was

⁽¹⁶⁾ Spirobacillene B (2): pale-yellow oil; $[α]_D^{25}$ –6.5 (*c* 0.05, CH₃CN); CD (CH₃CN) $λ_{max}$ (Δε) 324 (+4.06), 313 (–1.66), 234 (+4.04), 223 (–9.88) nm; UV (CH₃CN) $λ_{max}$ (log ε) 322 (3.90), 219 (4.08), 194 (4.19) nm; IR (film) v_{max} 3361, 2359, 1698, 1607, 1384, 835, 759 cm⁻¹; ¹H and ¹³C NMR spectra, see Table S2 in the Supporting Information; HRFABMS[M+H]⁺ *m*/*z* 292.0978 (calcd for C₁₈H₁₄NO₃, 292.0974).

⁽¹⁷⁾ Wilds, A. L.; Beck, L. W.; Close, W. J.; Djerassi, C.; Johnson, J. A., Jr.; Johnson, T. L.; Shunk, C. H. *J. Am. Chem. Soc.* **1947**, *69*, 1985–1994.

⁽¹⁸⁾ Haack, R. A.; Beck, K. R. Tetrahedron Lett. 1989, 30, 1605-1608.

⁽¹⁹⁾ Takagi, Y.; Mori, K. J. Braz. Chem. Soc. 2000, 11, 578–583.

⁽²⁰⁾ Zi, J.-C.; Lin, S.; Zhu, C.-G.; Yang, Y.-C.; Shi, J.-G. J. Asian Nat. Prod. Res. 2010, 12, 477–484.

⁽²¹⁾ Miles, D. W.; Eyring, H. Proc. Natl. Acad. Sci. U.S.A. 1973, 70, 3754–3758.

⁽²²⁾ Dong, J.-G.; Bornmann, W.; Nakanishi, K.; Berova, N. Phytochemistry 1995, 40, 1821–1824.

^{(23) (}a) Park, M. H.; Suh, D.-Y.; Han, B. H. *Phytochemistry* 1996, 43, 701–704.
(b) Liu, H.-W.; Nakanishi, K. J. Am. Chem. Soc. 1982, 104, 1178–1185.
(c) Hosoi, S.; Kamiya, M.; Ohta, T. Org. Lett. 2001, 3, 3659–3662.
(d) Harada, N.; Nakanishi, K.; Tatsuoka, S. J. Am. Chem. Soc. 1969, 91, 5896–5898.

determined to be (S)-3-hydroxy-2-(4-hydroxyphenyl)-spiro(cyclopentene-1,3'-indole)-4-one.



Figure 3. (a) CD (-) and UV (-) spectra of 2. (b) Sign of exciton chirality of 2.

To the best of our knowledge, spirobacillenes are a new type of naturally occurring indole alkaloid. The structure of 1 contains a single C-C bond between a spiro-[4.5]decane moiety and an adjacent indole ring, which is a novel skeleton for a natural product. The only report of spiro[4.5]decane natural products having originated from a microorganism involved the isolation of a bicarbonyl structure, spirodionic acid.¹⁵ In addition, although the carbon backbone of 2 has been previously reported as an intermediate in the synthesis of carbazole alkaloids,²⁴ the fact that a highly functionalized spiro-cyclopentenone that contains this carbon backbone has a natural origin is intriguing. The proposed biosynthetic pathway of spirobacillenes is shown in Scheme S1 (see Supporting Information (SI)). A time-course analysis for the metabolite profiles of L. fusiformis KMC003 resulted in the optimized

production of 3 in an 18 h broth culture of L. fusiformis KMC003, which nearly disappeared after 24 h of culturing (see Figure S3, SI). Notably, the production of 1 and 2 was maximized after the disappearance of 3, which suggested that 3 may be a key precursor for the biosynthesis of 1 and 2. Therefore, we presume that 3 could be the biosynthetic origin of the spirobacillenes. The biosynthesis of spirobacillenes most likely arises from subsequent decarboxylation and oxidation of 3a, which was proposed for the biosynthesis of the cyanobacterial metabolite. scytonemin.²⁵ The production of **3** could then undergo a keto-enol tautomerism and cyclization derived from the intramolecular reaction to afford the cyclopentenone ring and spiroaminal. The unique spiro[4.5]decane system in 1 may be generated from the divergent oxidative coupling of the carbanion from precursor 3. Compounds 1-3 were tested for their antimicrobial effects against 11 pathogenic strains. 1 and 2 exhibited weak activity with an MIC value greater than 50 μ g/mL, whereas 3 exhibited moderate antimicrobial activity against three strains, including Micrococcus luteus, Enterococcus hirae, and Staphylococcus aureus [MIC of 3: 3.13, 3.13, 12.5 µg/mL, respectively] (see Table S1). In addition, 1 and 2 were tested for inhibition against NO and ROS production in the LPS-induced RAW 264.7 macrophage cell line. The cytotoxic effect of 1 and 2 in RAW264.7 cells was evaluated by a CCK-8 assay. Compounds 1 and 2 did not affect cell viability for 48 h at concentrations up to 50 μ M. For following experiments, $2.5-50 \mu M$ concentrations of 1 and 2 were used. Compound 1 exhibited a weak inhibitory effect against the production of NO and ROS with IC50 values of 39 and 43 μ M, respectively, whereas 2 was found to be inactive.

Acknowledgment. The work was supported by the Global Leading Technology Program of the Office of Strategic R&D Planning (10039303) funded by the Ministry of Knowledge Economy, Republic of Korea.

Supporting Information Available. Experimental procedure, HPLC trace, NMR spectral data of 1-4, X-ray crystallographic data, and possible biosynthetic pathway of 1 and 2. This material is available free of charge via the Internet at http://pubs.acs.org.

⁽²⁴⁾ Beccalli, E. M.; Marchesini, A.; Pilati, T. J. Chem. Soc., Perkin Trans. 1 1994, 579–587.

⁽²⁵⁾ Balskus, E. P.; Walsh, C. T. J. Am. Chem. Soc. 2008, 130, 15260–15261.

The authors declare no competing financial interest.